

1A

A new, rapid and precise method for the quantitative separation of copper from bismuth or from bismuth antimony, and tin. G. Spacu and Irigina Pirtea (Univ. Bucharest, Rumania). *Anal. Rep. Populare Romaniae* (Univ. Stud., Ser.: Med., Phys., Chem. 2, 611, 1960) (French summary).—To the soln. contg. Cu and Sb ions add 0.75 g tartaric acid for each 0.1–0.3 g Sb. Dil. to 70–80 ml. add 2.5–3 ml. pyridine and 0.5 g. solut. NH_4SCN gradually while stirring. A green ppt. of $[\text{CuPy}_2(\text{SCN})_2]$ is formed immediately. When the ppt. has settled, filter through a filter crucible A. Wash the ppt. with a soln. of 0.75 g. NH_4SCN + 0.75 g. tartaric acid + 2.5 ml. pyridine + 2.5 ml. H_2O and then 7–8 times with 2–3 ml. of a soln. contg. 0.13 g. NH_4SCN + 3 ml. pyridine + 40 ml. H_2O in 20 ml. 90% EtOH . Finally wash with abs. EtOH and Et_2O contg. a small amt. of pyridine and dry for 20 min. in vacuo at room temp. If Bi, Sb, and Sn are present in the Cu alloy or mineral, treat with hot concd. HCl , introduce a few ml. H_2O , dropwise, heat until all metals are dissolved and the excess HCl removed. Add 0.75 g. tartaric acid, dilute to 75 ml., and add 5 ml. pyridine and 0.5 g. NH_4SCN with stirring. Wash as described above.

Gerhard Aufleger

1ST AND 2ND COVER										3RD AND 4TH COVER									
CONTENTS AND COMPUTER CODE																			
②										A-1									
Reprecipitation of Ammonia of hydrochloride acid in the presence of hydrochloric acid: G. G. Lerner and J. J. Jones, <i>Anal. Chem. Soc. Boston</i> 1950, 22, 1000. The mixture of chloride and ammonia is used, completely, by AgNO₃, after centrifugation. The washed ppt. is estimated with 0.1 N. KBr, 0.1 N. of 0.1% KBr is added, the solution adjusted with HNO₃, and filtered. Any chloride in the original ppt. now appears as KCl in the filtrate (AgCl + KBr + HCl + AgBr). If on the addition of AgNO₃, the ppt. formed is pure white, and therefore contains no AgBr, the original ppt. must have contained at least 1 mg. of AgCl. The operation may be repeated with successive amounts of 1 cc. of the KBr solution until a yellow ppt. is obtained; from the amount of KBr used the Cl content of the original solution can be calculated. A. B. M.																			
ADD-514 METALLURGICAL LITERATURE CLASSIFICATION																			
FROM SYNDICATE										FROM LIBRARY									
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PHASE I BOOK EXPLOITATION

375

Katsnel'son, Genrikh Mayorovich; Saf'yan, Matvey Matveyevich;
Chekmarev, Aleksandr Petrovich; Mal'yy, Georgiy Ivanovich

Prokatka tolstykh listov s povyshennoy tochnost'yu (Rolling of
Steel Plate to Close Limits) Moscow, Metallurgizdat, 1957.
125 p. 4,000 copies printed.

Ed. (title page): Chekmarev, A. P., Active Member, Ukrainian
Academy of Sciences, Doctor, Professor; Ed. (inside book):
Pirskiy, P. N.; Ed. of Publishing House: Valov, N. A.;
Tech. Ed.: Karasev, A. I.

PURPOSE: This book is intended for engineers and technicians in
rolling mills. It can also serve as a manual for
researchers and students of vuzes.

COVERAGE: The book deals with the hot rolling of steel plate to
close limits on a three-high Lauth mill. Various factors
affecting the precision of rolled plate are discussed.
The rolling of plate is subject to variables such as:
temperature of metal, mill spring, roll design, and other
characteristics inherent in the material and equipment.

Card 1/3

Rolling of Steel Plate to Close Limits

375

The author investigates each of these problems and advances various solutions. There are numerous diagrams and formulae. 6 Soviet references.

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AVAILABLE: Library of Congress (TS 360.C45)

Card 3/3

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(A) L 65136-65 EWT(d)

ACCESSION NR: AP5021631

UR/0286/65/000/013/0112/0112

AUTHORS: Vasilenko, N. T.; Vasil'yev, Yu. P.; Orlov, Yu. V.; Pirskiy, P. K.

TITLE: A hinge for connecting pontoons. Class 65, No. 172643

SOURCE: Byulleten' izobreteniy i tovarnykh znakov, no. 13, 1965, 112

TOPIC TAGS: pontoon, mechanical fastener

ABSTRACT: This Author Certificate presents a hinge for connecting pontoons, made in the form of two brackets fixed to the flanges of adjacent pontoons and joined by an axle (see Fig. 1 on the Enclosure). To facilitate and expedite joining pontoons for floating in waves, the axle of the hinge is fixed on two radially-spherical bearings pressed into the bracket. In its central portion, the cross section of the axle is square. This square portion enters into a slot of the other bracket which also has a slot perpendicular to the first one. The second slot forms a seat for a wedge which locks the hinge when the pontoons are connected. Orig. art. has: 1 figure.

ASSOCIATION: none

SUBMITTED: 31Mar64

NO REF SOV: 000

Card 1/2

ENCL: 01
OTHER: 000

SUB CODE: IE

L 65136-65

ACCESSION NR: AF5021631

ENCLOSURE: 01

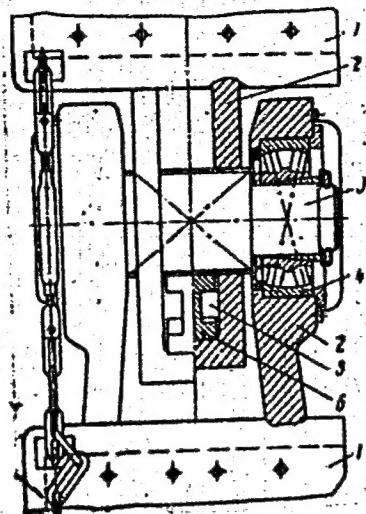


Fig. 1. 1- pontoons; 2- brackets; 3- hinge axis; 4- radially-spherical bearings; 5- slot; 6- wedge

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FIRSTTEL, Jerzy, inz.

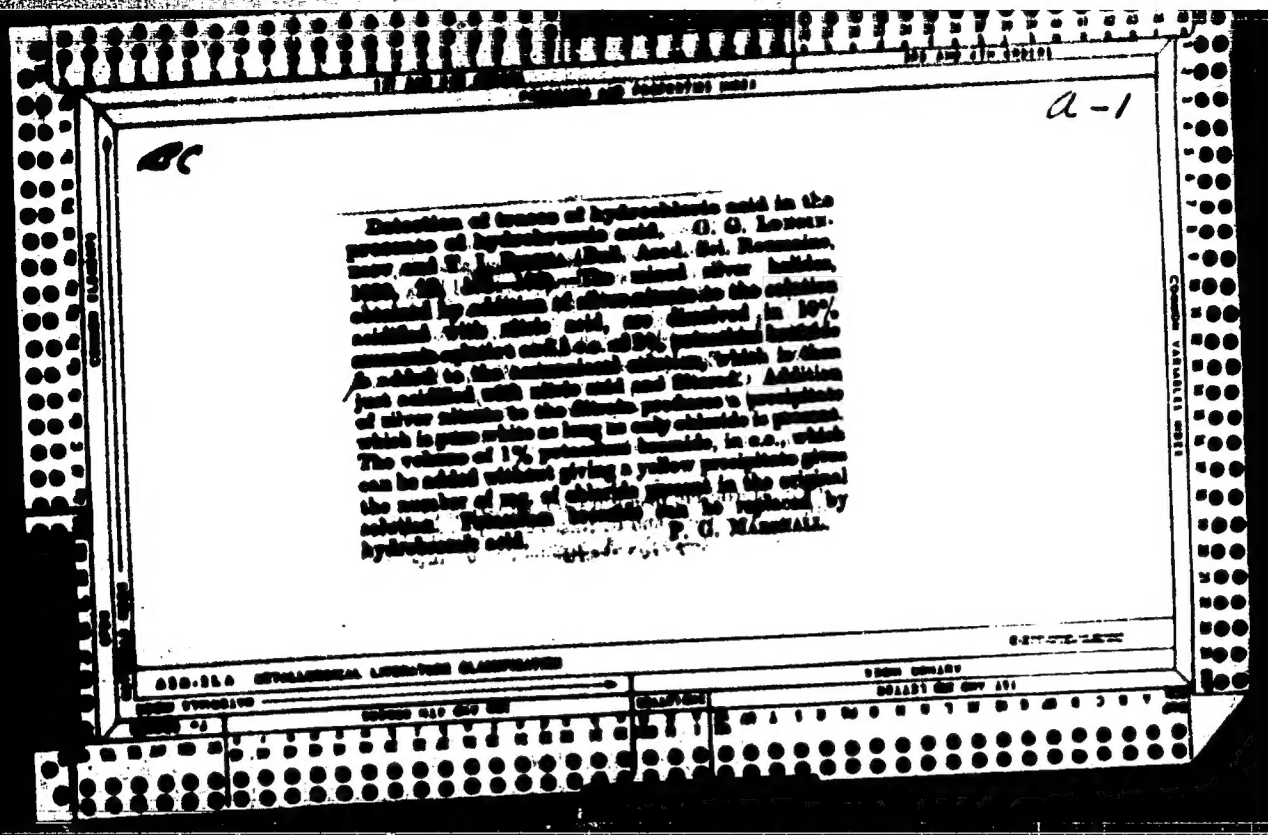
Behavior of voltage transformers in 10 kv networks with insulated neutral point. Energetyka Pol. 17 no. 1209-211 31 '63.

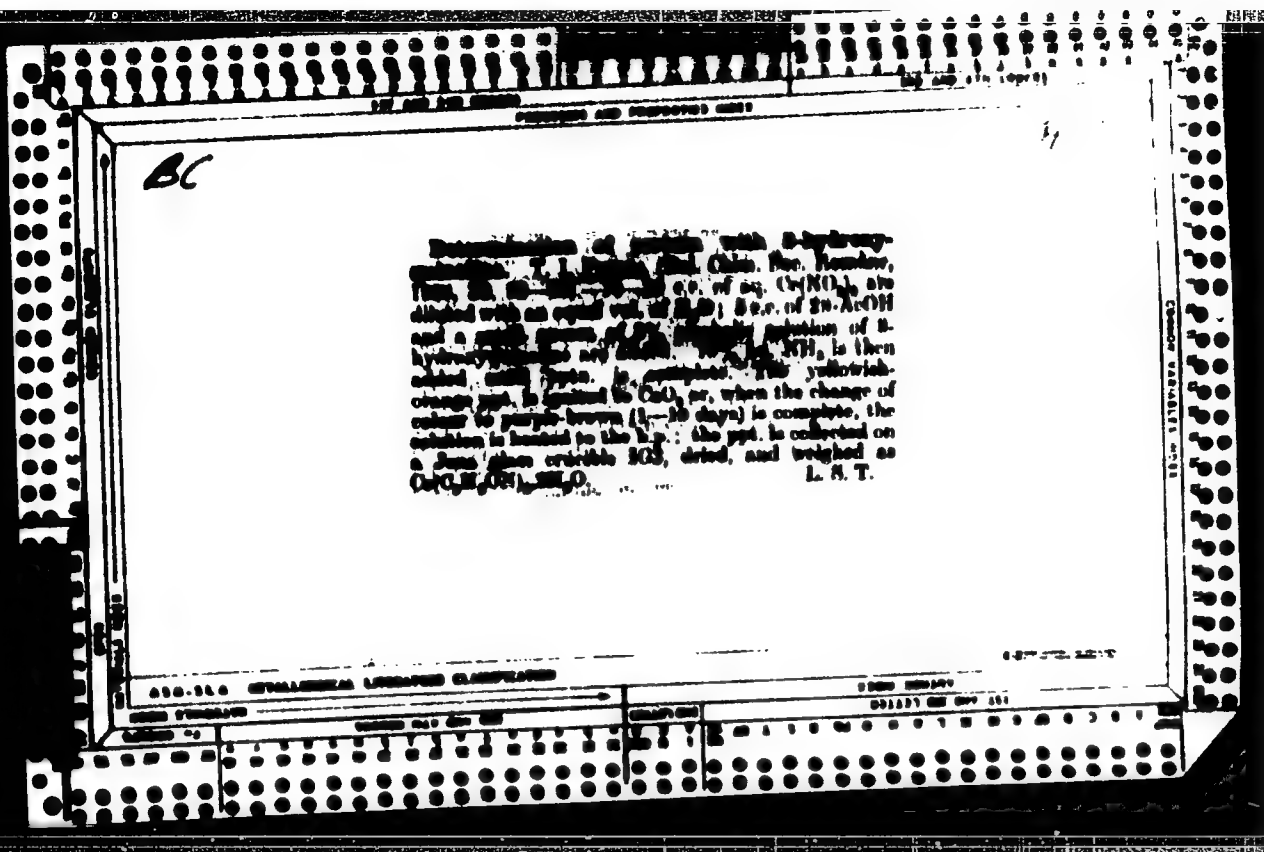
1. Zaklad Energetyczny, Gdansk.

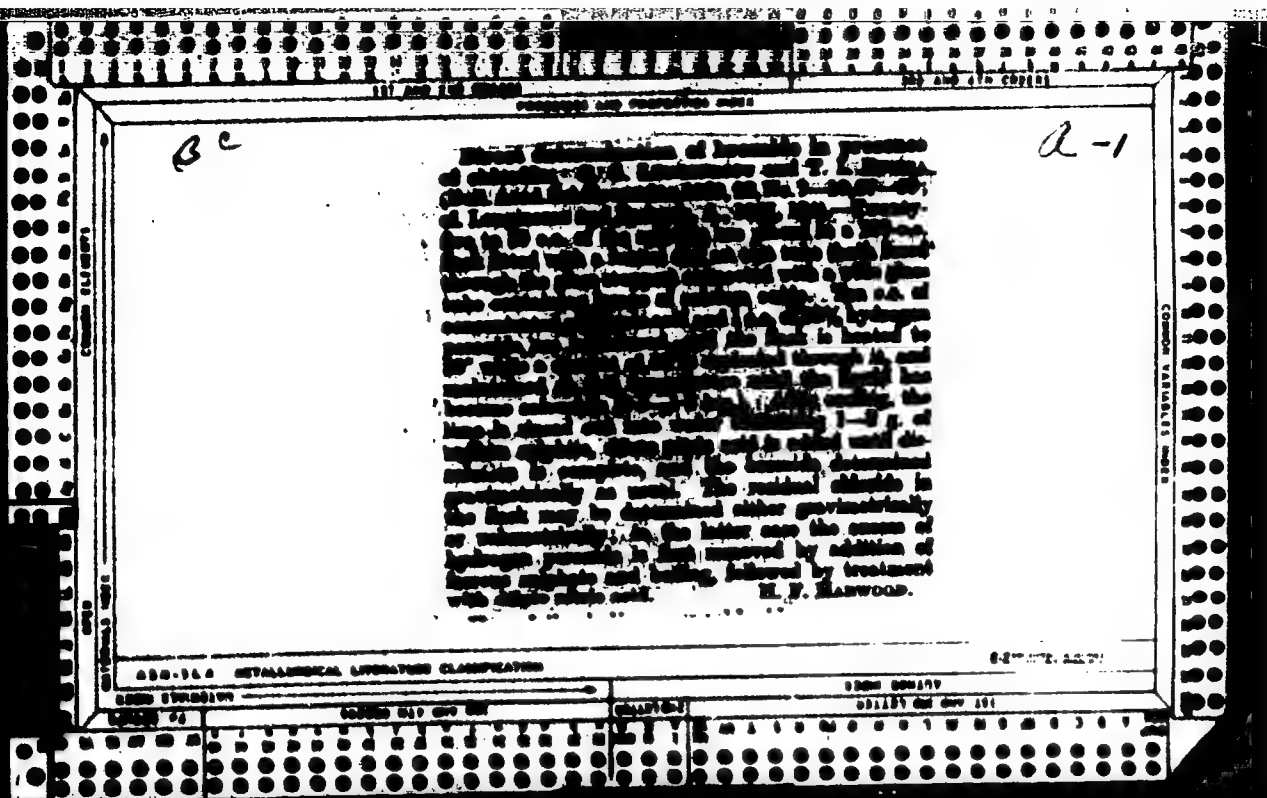
MIELINSKI, Jan, mgr inż.; PIROTEL, Jerzy, inż.; KUSOWSKI,
Włodzisław

Methods of decreasing the number of possible 15 kv cable
faults. Energetyka Pol 18 no. 2: 44-45-48 F '64.

1. Zakład Energetyczny, Gdansk.







ca

7

Determination of cerium with 8-hydroxyquinoline 1b
 1. Putra. *Bull. Chim. Soc. Romania* (1937-8) (in German). To 20-40 ml of a neutral soln
 contg up to 0.10 g of Ce^{+4} , add 5 ml of 2 N AcOH
 and a slight excess of a 3% soln of oxine in EtOH. No
 ppt will form. Now add from a pipet 10% NH_4 soln
 until the Ce is pptd. The ppt has a yellow-orange
 color which slowly becomes brownish purple. The latter
 compound has the more definite constitution. After the
 change in color is complete, heat to boiling, filter through
 a glass filtering crucible, wash with hot water, dry and
 weigh. The ppt has the formula $Ce(C_{15}H_8O_2N_2)_2 \cdot 2H_2O$
 obviously an oxidation of the Ce^{+3} has taken place.
 The ppt contains only 18.3% Ce . The results obtained
 in 14 expts. in which the ppt was weighed were excellent,
 but it takes so long to complete the transformation to
 ceric salt that it is more practicable to ignite the ppt and
 weigh the residual CeO_2 . Good results were obtained in
 10 analyses. W. I. H.

✓

Detection of nitric acid by the formation of fuchsin. G. G. LONGINOV AND IN
 I. P. IONINA. *Russ. chem. rev.* 30, Nov 1961, pp. 1041-1042. The test depends
 upon the formation of $C_6H_5NO_2$ by treatment with a little benzene and concentrated H_2SO_4 ,
 reducing the nitrobenzene to aniline and making the aniline react with $HgCl_2$ to form
 fuchsin. To a little of the substance in a test tube add 2-3 drops of benzene and 1 cc
 of concd. H_2SO_4 . With the temp. of the mixt. approx. 30° add a little water and intro-
 duce powdered Zn little by little. After the $C_6H_5NO_2$ has been reduced to $C_6H_5NH_2$,
 divide the soln. into 2 parts and test one portion for aniline by the reaction with $CaCl_2$.
 To the other portion add a little solid $HgCl_2$ and evaporate just to dryness under the hood.
 Continue heating carefully until a reddish green coloration appears on the sides of the
 tube. Cool, add a little HCl and the presence of fuchsin will be shown by the reddish
 violet color of the soln. W. L. H.

MA

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The Quantitative Estimation of Cerium with Ortho-oxoquinoline. Th. I. ... (In German). Dilute to 20 ml. of a 0.1% solution with the same amount of water, acidify with 5% of 2N HCl, and add a slight excess of 3% alcoholic solution of ceric ammonium sulphate drop by drop until precipitation is complete. Allow the orange-yellow precipitate to settle and to remain until it has become entirely purple brown in colour, then bring to the boil and filter through a glass filter, wash with 10% HCl, dry, and weigh. The dried precipitate has the composition $Ce_2(H_2O)_4 \cdot 2H_2O$. The colour change takes place in 1-10 days. A much quicker way with the same accuracy is to filter the hot precipitate immediately after formation through a quantitative filter, wash with hot water, and ignite to Ce_2O_3 .

1-143

7

A new rapid and precise method for the determination of aluminum. G. Sauer and Th. I. Petre (Univ. Bucharest, Romania). *Acad. Rep. Populare Romane, Bul. Stint.* Ser. Mat., Fiz., Chim. 2, 619 (1969) (French summary).
Treat a soln. contg. 0.001-0.03 g. Al with an excess (2-8 ml.) of a 10-15% soln. of Na mercaptobenzothiazole, C₆H₄(SH)SSNa. After stirring, filter through a porcelain filter crucible A or A₂ or Jena crucible K₁. Transfer all the ppt. to the crucible with a soln. of 0.1 g. reagent in 100 ml. H₂O. Wash the ppt. 3-4 times with 2-3 ml. portions of H₂O and dry at 105-110° for 30-45 min., then weigh. Preps. of the reagent: Treat mercaptobenzothiazole with a 5% soln. of NaOH. Use a little less than the stoichiometric amt. of NaOH and remove the excess mercaptobenzothiazole by filtering. The obtained soln. of C₆H₄(SH)SSNa has a pH of 8.
Gerhard Aufleger

New method for the gravimetric determination of thorium
 G. Nemes and Th. I. Pirtes (Univ. Bucharest, Rumania)
Acad. Rep. Populare Romane, Bul. Stint. Ser. Mat., Fiz.
Chim. 2 (1961) 76 (1961) French summary To 5.0 ml
 of a soln. contg. 0.02-0.2 g. Th. add 2.0 ml. of the
 soln. of metapictobenzothiazide soln. while stirring.
 A white ppt. is formed instantaneously. After stirring for 5
 min. filter with a filter crucible A or Jena RG, wash with 50
 ml. of a soln. contg. 1-1.5 ml. reagent in 100 ml. H₂O.
 then 4-6 times with 2 ml. H₂O, dry at 105-110° for 30-45
 min. and weigh as (C₁₂H₈N₄S₂Th). It is important that a
 large excess of reagent is used. To prep. the
 reagent see preceding abstract. (Richard Aufleger)

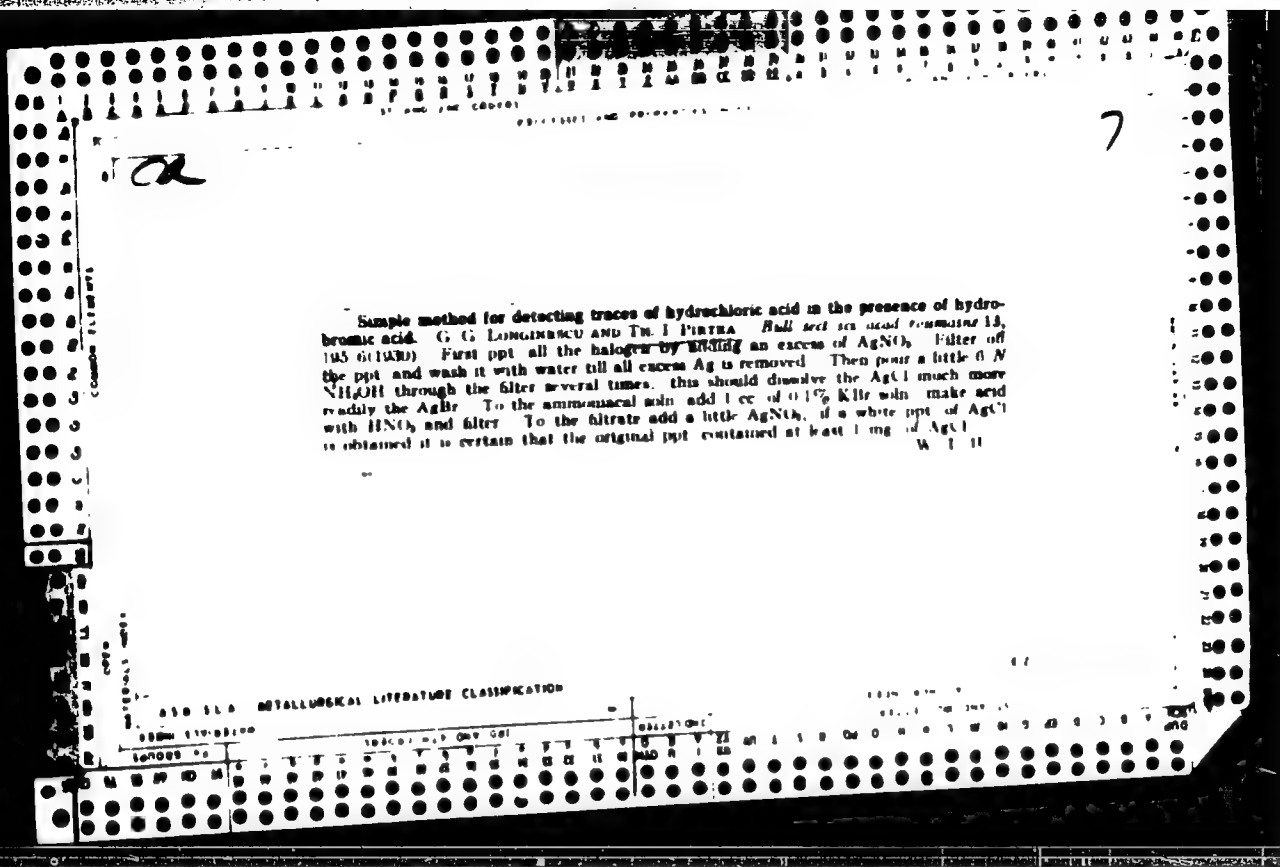
CA

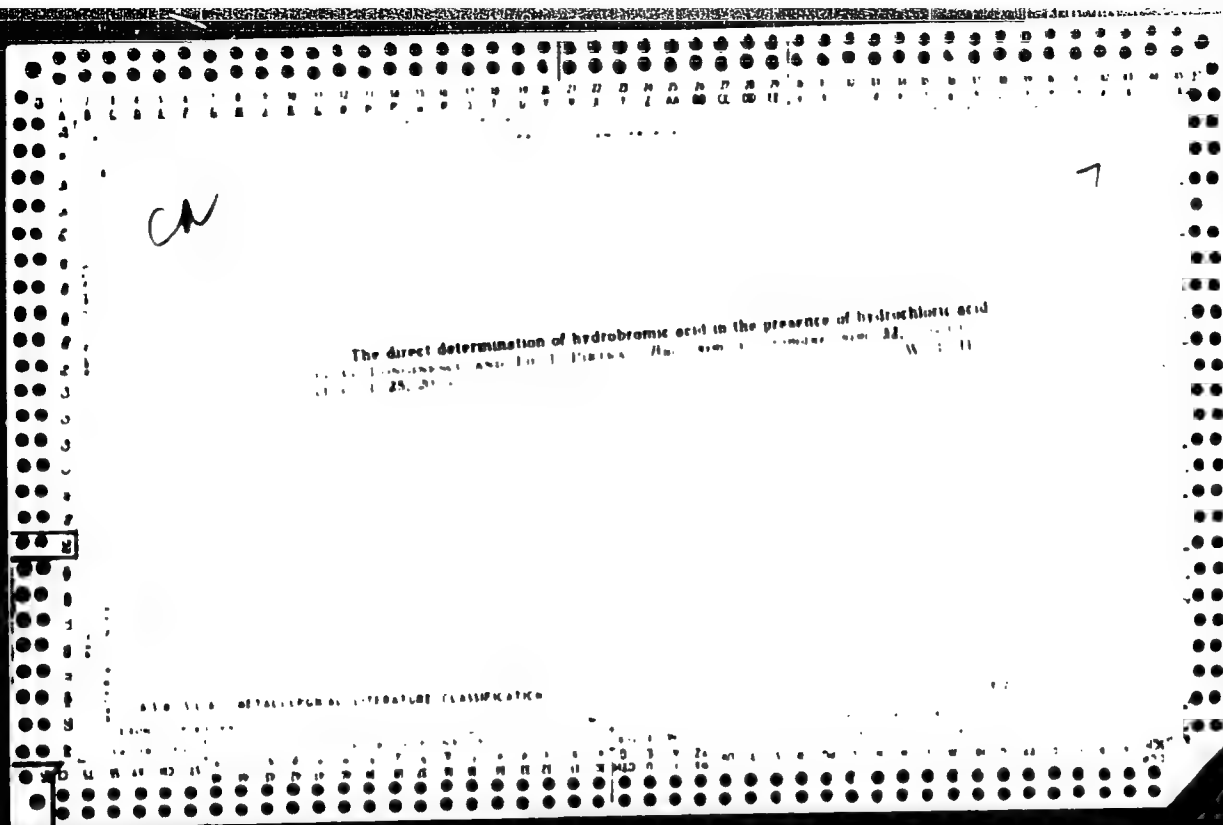
New method for the gravimetric determination of thorium
 C. Spacu and Th. I. Pirtea (Univ. Bucharest, Romania)
Acad. Rep. Populare ROM. Ser. Mat. Fiz. Chim. 2 (1960) French summary. To 5.20 mg
 of a soln contg 0.02-0.2 g Th NO_3 , add 2-10 ml of the
 Na salt of mercaptobenzothiazole soln while stirring.
 A white ppt is formed instantaneously. After stirring for 5
 min, filter with a filter crucible A or B, wash with 50
 ml of a soln contg 1-1.5 ml reagent in 100 ml H_2O ,
 then 4-6 times with 2 ml H_2O , dry at 105-110° for 30-45
 min, and weigh as $(\text{C}_6\text{H}_4\text{NS}_2)_2\text{Th}$. It is important that a
 large excess (3-4 times) of reagent is used. To prep. the
 reagent see preceding abstract. Gerhard Aufleger

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ca

Determination of hydrochloric acid in the presence of hydrobromic and hydriodic acids. G. G. LORINGHAM AND J. H. L. HARRIS. *Anal. Chem.* 28, 1045 (1956). The method described depends upon first precipitating the silver as AgCl with AgNO_3 in dil. HNO_3 and weighing the ppt. This gives the wt. of AgCl together with AgBr and AgI . In a second portion of the same soln. the ppt. is kept as far as possible in the beaker and washed by decantation to remove all excess Ag^+ . The ppt. is then heated with HNO_3 of 7-8% NH_4OH soln. and treated with a soln. of KIO_3 in HNO_3 , whereby any chloride of Ag is converted into AgIO_3 . This ppt. is also weighed finally, when all 3 halides are present, a similar treatment is given to the ppt. except that KI is added after the treatment with NH_4OH soln. In this case the total halide content is determined as AgI . The purpose of the NH_4OH treatment is to dissolve all the AgCl and part of the AgBr , whereby the KIO_3 or KI soln. becomes more effective in accomplishing the desired conversion.

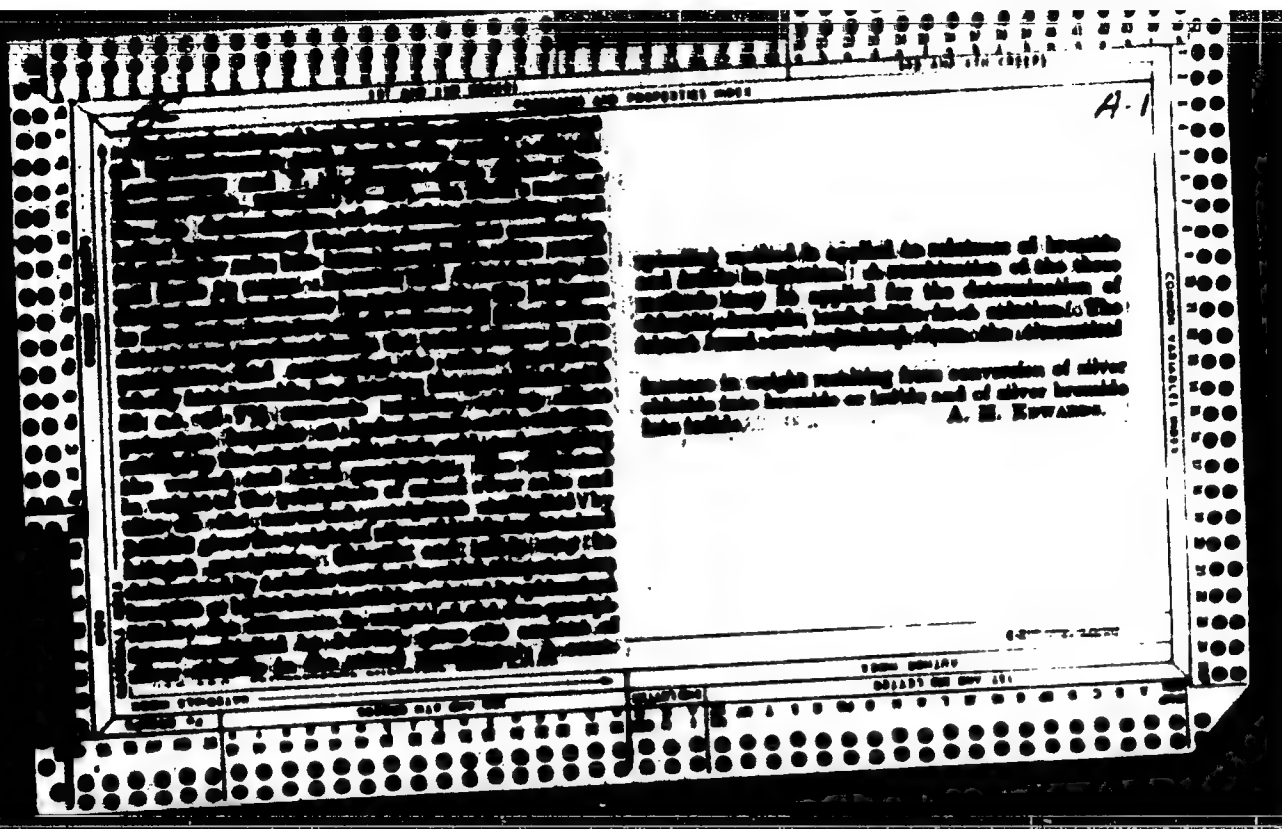




Direct determination of hydrobromic acid in the presence of hydrochloric acid
 C. G. LONGMANS AND T. J. POTA. *Anal. Chem.* 13, Nov. 7, 1941, 1491-1492. Introduce 20 ml. of the acid into a 250 cc. glass stoppered flask which is provided with a tap funnel leading to the bottom of the flask and with a delivery tube both entering the flask through the ground glass stopper. The exit tube leads to a tube of difficultly fusible glass 0.6-0.7 cm. in diam. and about 15 cm. long filled with calcined lime or magnesia. With the former it is necessary to heat the lime to dull redness. Add to the contents of the flask through the tap funnel, 10 cc. of concd. H_2SO_4 and 1 cc. of perhydrol. By means of suction, pass a current of air through the app. Heat the contents of the flask on the water bath to about 60°. The current of air carries along the H_2 liberated by the oxidation of the HBr and the HCl is retained by the CaO or MgO . Continue heating and passing the air until the liquid in the flask is colorless, which takes 0.5-1.5 hrs. Distill the Cl_2 in the flask gravimetrically as $AgCl$ or by KCN titration after the addition of excess Ag . Empty the absorbent into 200 cc. of water and add 1-2 g. of Na_2SO_3 to reduce any bromate that may have formed. Wash out the tube with HNO_3 , filter off any impurity and distill the H_2 as AgH_2 . The numerous results given show that the method is accurate. W. T. H.

CA

A new rapid and precise method for the determination of
aluminum. L. Sporn and Th. J. Pater, J. Inorg. Nucl. Chem.
Romania, 1968, 10, 2, 101-102. In a 100 ml solution of
the sample, add 2 ml of 10% sodium hydroxide solution.
Treat a 100 ml solution of Na metacaptodisulfide with
of a 10% solution of Na metacaptodisulfide. After
stirring, filter through a porcelain filter.
crucible. After stirring, filter through a porcelain filter.
to the crucible with a solution of 0.1 g reagent in 100 ml H₂O and
wash the ppt 4 times with 2 ml portions of H₂O. Dry at 105°C for 30 min, then weigh. Prepare the
reagent. Treat metacaptodisulfide with a solution of
NaOH. Use a little less than the stoichiometric amount of
filtering. The obtained solution of C₁₂H₁₀Na has a pH of 8.
Richard Aulinger



PIRTEA, D.

2093. New gravimetric method for the rapid estimation of copper. D. Pirtea, E. Buzatu and S. Zelig. *Stud. Cercet. Chim., Bucharest, 1960, 1 (3-4), 233-236.*—Copper is pptd. quant. as $(CuPy)_2(ClO_4)_2$ by the addition of pyridine and solid $NaClO_4$ at room temp. After filtration, the ppt. is finally washed with ether and is dried in a vacuum desiccator. The analysis can be performed in < 15 min.
J. H. WATSON

ha
107

PIRTEA, D.

2512. New gravimetric methods for the rapid estimation of the elements nickel, cobalt and cadmium. D. Pirtea, G. Dumitrescu and N. Alecu, *Stud. Cercet. Chim., Bucharest*, 1955, 8 (12-5), 237-242. Nickel, Co and Cd are each pptd. quant. as the complex $[MPy_3][Cr_2O_4]$ by the addition of pyridine and $K_2Cr_2O_7$ soln. at room temp. After filtration, the ppt. are washed with ethanol and with ether, and are dried in a vacuum desiccator. Each analysis can be performed in 45 to 60 min.

J. H. WATON

fra
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PURTER, DESPINA.

Handwritten: 11/11/51
The separation and gravimetric determination of copper in the presence of lead or stannous or of both of these metals. Cu, Spacu and Despina Putea. *Comm. Acad. Rep. Populare Română* 3, 71-72 (1953). The method of Cu as and Dick (C.A. 21, 2633), consisting in the pptn. of Cu as a pyridine complex (CupyrSCN₂), was used for the sepn. of Cu from Fe⁺⁺⁺ and Al⁺⁺⁺, which were kept in soln. by tartaric acid (cf. C.A. 45, 7812a). The Fe⁺⁺⁺ and Al⁺⁺⁺ ions were then pptd. from the filtrate with 8-hydroxyquinoline. The method was rapid and precise. P. Keftest

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Handwritten: PM

LIPTA, L.; DUMITRISCU, G.; PELICAN, N.

New gravimetric methods for rapid determination of the elements
nickel, cobalt, and cadmium, p. 238. Academia Republicii Populare Romane.
STUDII DE TERCELEA DE CLASIE. Bucuresti. Vol. 3, no. 34, July/Dec. 1965.

So. Eng. European Accessions List Vol. 5, No. 9 September, 1961

PIRTEA, D. ; ALBESC", I.

The macro-and microgravimetric method of determining the mercury in substances for the protection of plants. p. 137.

STUDII SI CERCETARI DE CHIMIE. Bucuresti, Rumania
Vol. 7, No. 1, 1958.

Monthly List of East European Accession (EFAI). LC, Vol. F, No. 9, Sept. 1958
Uncl.

L 34899-66 EMP(t)/ETI IJF(c) LID
~~ACC NR~~ AP6026618

SOURCE CODE: RU/0003/65/016/005/0288/0289

AUTHOR: Pirtea, T. I.

ORG: Laboratory for Analytical Chemistry, Faculty of Chemistry, Bucharest University
(Laboratorul de chimie analitica, Facultata de chimie, Universitatea Bucuresti)

TITLE: New methods for the gravimetric determination of cadmium and lead

SOURCE: Revista de chimie, v. 16, no. 5, 1965, 288-289

TOPIC TAGS: cadmium, lead, chemical precipitation, quantitative analysis

ABSTRACT: The author describes a new gravimetric method for the determination of cadmium and lead through precipitation with aqueous solutions of α, α' -dipyridyl hydrochloride or α, α' -dipyridyl nitrate in the presence of ammonium thiocyanate. This method can be used in the presence of the elements Na, K, Ca, Sr, Ba, Mg, Be, Al, Cr, Sc, and NH_4^+ ions. Orig. art. has: 2 figures and 2 tables. [Based on author's Eng. abst.] [JPRS]

SUB CODE: 07 / SUBM DATE: none / ORIG REF: 003 / OTH REF: 008

Card 1/1

UDC: 546.48.04:546.815.04:545.12

PIRTEA, Th. I.

A new method for the gravimetric determination of manganese.
Rev chimie Min petr 15 no.10:635-636 C '64.

1. Laboratory of Analytical Chemistry, Faculty of Chemistry
Bucharest University.

L 64818-65 EWP(t)/EWP(b) IJP(o) JD

ACCESSION NR: AP5023231

HJ/0003/64/015/010/0635/0636

AUTHOR: Pirtea, Th. I.

TITLE: New method for the gravimetric determination of manganese 21

SOURCE: Revista de chimie, v. 15, no. 10, 1964, 635-636

TOPIC TAGS: manganese, gravimetric analysis

ABSTRACT:

The author describes a rapid and reliable method for the gravimetric determination of manganese by precipitating the Mn^{++} ions with α, α' -dipyridyl hydrochloride in the presence of ammonium thiocyanate. The method may be used for the determination of manganese in the presence of magnesium, antimony, and the alkali and alkali earth elements if tartaric acid is used as a screening agent. Orig. art. has: 1 table, 1 figure.

Card 1/2

L 64818-65

ACCESSION NR: AP5023231

ASSOCIATION: Laboratorul de chimie analitica, Facultatea de chimie, Universitatea
Bucuresti (Laboratory of Analytical Chemistry, Faculty of Chemistry, University
of Bucharest)

SUBMITTED: 00

ENCL: 00

SUB CODE: IC, DC

NR REF SOV: 000

OTHER: 009

JPRS

MJR
Card 2/2

FIRTEA, Th 1.

Distr: LE2c

A new microgravimetric method for the determination of aluminum. Th 1 Firtea and Georgeta Mihail (Univ. C. I. P. I. Bucharest, Romania). *Analele Univ. C. I. P. I. Bucharest Ser. stint. nat.* 15, 83-6 (1957).—This method uses the Na salt of mercaptobenzothiazole as the pptg. agent. The Al must be in a neutral soln. as the chloride, nitrate, or sulfate salt. Free acids and salts of acetic, tartaric, and oxalic acid must be absent since they will render the soln. acidic and will ppt. the reagent. To a 0.5 ml.-5 ml. soln. of the unknown 0.5-1 ml. of the reagent is added contg. 0.1-0.15 g. of Na mercaptobenzothiazole (1) per ml. of water. The pptn. is immediate and complete at room temp. After some shaking the soln. and ppt. are filtered and washed with a soln. contg. 0.5 g. of I per 100 ml. of water. The ppt. is then dried for 30 min. at 110-15°. It is weighed as the anhydride. This detn. can be performed in the presence of alkali metals and Mg. *Recueil*

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(11)

A new method for the microgravimetric determination of cobalt. Separation and determination in the presence of nickel. Th. I. Pirica and Polly Nucan. *Analyst* 1967, 92, 101-102. *C.I. Parson, Document, Ser. 1111, vol. 16, 71-4*

(1967).—A new method is given for the microgravimetric detn. of Co as the complex $[\text{Co}(\text{NH}_3)_6][\text{Co}(\text{NO}_2)_6]$ (I). By treating a soln. of Co^{++} with NaNO_2 , AcOH , and an excess of NaNO_2 , the anionic complex $[\text{Co}(\text{NO}_2)_6]^{--}$ or $\text{Na}_2[\text{Co}(\text{NO}_2)_6]$ is obtained. When this soln. is treated at room temp. with an aq. satd. soln. of $[\text{Co}(\text{NH}_3)_6]\text{Cl}_2$, I is pptd. as yellow insol. crystals. The ppt. is filtered after a few min., washed with alc. and abs. ether, dried for 10 min. in a vacuum desiccator, and weighed. The method is precise and rapid because of the insol. and high mol. wt. of I. Co can also be detd. in the presence of Ni with this method. C. Heitner-Wirgin—

Distr: 4E2c

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PIRTEA) 1.

Distr: LE2c(j)/LE2c

A new method for the potentiometric determination of silver in the presence of other elements. P. Spacu and Th. T. Pirtea (Univ. C. I. Parhon, Bucharest, Romania). *Chim. Ind. (Milan)* 39: 1567-1568 (1957).—This new potentiometric method of titration uses as a reagent a 0.1N soln. of Na nitroprusside ($\text{Na}_2[\text{Fe}(\text{CN})_5\text{NO}] \cdot 2\text{H}_2\text{O}$). In order to cause the coagulation of the ppt. of Ag nitroprusside, 5-6 g. of NaNO_3 is added previously to the soln. the total vol. of which should be around 100-150 ml. This allows for a better break in the potential curve at the end point. The potential of the system before the titration is around 400 mv. and the inflection point at 310 mv. with respect to the normal calomel electrode. A Ag wire was used as indicating electrode. The soln. has to be agitated during the whole titration. The presence of Zn and Pb does not disturb the titration since they do not form ppts. with the reagent. In the case of Cu and Cd they have to be masked with Complexon III or the Na salt of ethylenediaminetetraacetic acid. In such cases approx. 1 g. of the complexon is added. A. Berlin

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Pirtea T.I. I
RUMANIA/Analytical Chemistry - Analysis of Inorganic Substances.

Abs Jour : Ref Zhur - Khimiya, No 8, 1958, 24752

Author : Pirtea, Th.I., Strac, A.

Inst :

Title : New Micro-Gravimetric Method for Determination of Zinc.

Orig pub : Rev. chim., 1957, 8, No 6, 435-437

Abstract : Description of a method based on precipitation of Zn²⁺ with the Na-salt of mercaptobenzothiazole at room temperature. The precipitate is filtered off immediately, dried at 110° and weighed. The conversion factor is 0.1644.

Card 1/1

RUMANIA/Analytical Chemistry - Analysis of Inorganic Substances.

E-2

Abs Jour : Ref Zhur - Khimiya, No 8, 1958, 24742

Author : Spacu, P., Pirtea, Th.I.

Inst : -

Title : New Method of Potentiometric Determination of Silver in the Presence of Other Elements.

Orig Pub : An. Univ. "C.J. Iarhon". Ser. stiint. natur., 1957, N 15, 67-71

Abstract : Ag^+ is determined by potentiometric titration with 0.1 N solution of Na nitroprusside in a weakly acidic solution and in the presence of 5-6 g $NaNO_2$ which serves to coagulate the resulting colloidal $Ag_2Fe(CN)_6(NO)$. An Ag wire is used as the indicator electrode and a calomel electrode as the reference electrode. The presence in the solution being titrated of Pb^{2+} (up to 50 ml of 0.1 N solution) and of Zn^{2+} (up to 30 ml of 0.1 N solution) does not interfere with determination of Ag. In the presence of Cu^{2+}

Card 1/2

RUMANIA/Analytical Chemistry - Analysis of Inorganic Substances.

E-2

Abs Jour : Ref Zhur - Khimiya, No 8, 1958, 24742

(30 ml 0.1 N solution) and Cd^{2+} (5 ml 0.1 N solution). 0.8-1 g Complexon III are added to the solution being titrated in order to mask these ions. NO_2^- , CH_3COO^- , SO_4^{2-} do not interfere. Determination error does not exceed 2%.

Card 2/2

RUMANIA/Analytical Chemistry - Analysis of Inorganic Substances.

E-2

Abs Jour : Ref Zhur - Khimiya, No 8, 1958, 24761

Author : Pirtea, Th.I., Mihail Georgeta

Inst :

Title : New Micromethod of Gravimetric Determination of Aluminum

Orig Pub : An. Univ. "C.J. Parhon". Ser. stint. natur., 1957, No 15, 83-86

Abstract : On interaction of Al^{3+} with the Na-salt of mercapto-benzothiazole (I) in neutral or weakly acidic media (pH 6) there is formed a white crystalline precipitate of $Al(C_7H_4NS_2)_3$, which is practically insoluble in water and in excess of the reagent, and which is suitable for a gravimetric determination of Al. To 0.5-5 ml of the solution being analyzed, containing about 0.5 mg/ml of Al, are added 0.5-1 ml of 10-15% solution of I. The mixture is stirred, filtered and washed 2-3 times with 0.5% solution of I, and 2-3 times with water. The precipitate so obtained

Card 1/2

ROMANIA/Analytical Chemistry - Analysis of Inorganic Substances. E-2

Abs Jour : Ref Zhur - Khimiya, No 8, 1958, 24761

is dried at 110-115° and weighed. The conversion factor is 0.051307. Ions of alkali metals and Mg do not interfere with the determination. In the presence of free mineral acids, acetates, oxalates and tartrates, the I itself is precipitated, hence they must be removed beforehand. The described method is not inferior in accuracy to the 8-hydroxy-quinoline method.

Card 2/2

21

Plata, Th I.

Distr: 4E2c

27

/ Microgravimetric determination of thorium. Th. I.
~~Berlin and Georgeta Mihail~~ (Univ. C. I. Parhon, Bucharest,
 Romania). *Analele Univ. "C. I. Parhon" Bucuresti Ser.*
chim. vol. No. 12, 51-5 (1956).—Th is pptd. as $\text{Th}(\text{C}_4\text{H}_5\text{NS})_4$
 with a soln. of the Na salt of mercaptobenzothiazole. The
 pptn. is immediate at room temp., and the ppt. is white and
 cryst. It is filtered immediately and dried 30-60 min. at
 110-115°. This method can be used to det. Th in the
 presence of the salts formed by the alkali and by Mg with
 strong acids. Free acids must be absent from the Th soln.,
 as they will ppt. the pptg. agent. A. Berlin

Distr: LE2c

A new microgravimetric method for the determination of aluminum. Th. I. Purtes and George M. ...
C. I. Purtes, Bucharest, Romania. ...
Purtes, Bucharest. See abstract 15, 80, 81, 82, 83, 84, 85, 86, 87, 88, 89, 90, 91, 92, 93, 94, 95, 96, 97, 98, 99, 100, 101, 102, 103, 104, 105, 106, 107, 108, 109, 110, 111, 112, 113, 114, 115, 116, 117, 118, 119, 120, 121, 122, 123, 124, 125, 126, 127, 128, 129, 130, 131, 132, 133, 134, 135, 136, 137, 138, 139, 140, 141, 142, 143, 144, 145, 146, 147, 148, 149, 150, 151, 152, 153, 154, 155, 156, 157, 158, 159, 160, 161, 162, 163, 164, 165, 166, 167, 168, 169, 170, 171, 172, 173, 174, 175, 176, 177, 178, 179, 180, 181, 182, 183, 184, 185, 186, 187, 188, 189, 190, 191, 192, 193, 194, 195, 196, 197, 198, 199, 200, 201, 202, 203, 204, 205, 206, 207, 208, 209, 210, 211, 212, 213, 214, 215, 216, 217, 218, 219, 220, 221, 222, 223, 224, 225, 226, 227, 228, 229, 230, 231, 232, 233, 234, 235, 236, 237, 238, 239, 240, 241, 242, 243, 244, 245, 246, 247, 248, 249, 250, 251, 252, 253, 254, 255, 256, 257, 258, 259, 260, 261, 262, 263, 264, 265, 266, 267, 268, 269, 270, 271, 272, 273, 274, 275, 276, 277, 278, 279, 280, 281, 282, 283, 284, 285, 286, 287, 288, 289, 290, 291, 292, 293, 294, 295, 296, 297, 298, 299, 300, 301, 302, 303, 304, 305, 306, 307, 308, 309, 310, 311, 312, 313, 314, 315, 316, 317, 318, 319, 320, 321, 322, 323, 324, 325, 326, 327, 328, 329, 330, 331, 332, 333, 334, 335, 336, 337, 338, 339, 340, 341, 342, 343, 344, 345, 346, 347, 348, 349, 350, 351, 352, 353, 354, 355, 356, 357, 358, 359, 360, 361, 362, 363, 364, 365, 366, 367, 368, 369, 370, 371, 372, 373, 374, 375, 376, 377, 378, 379, 380, 381, 382, 383, 384, 385, 386, 387, 388, 389, 390, 391, 392, 393, 394, 395, 396, 397, 398, 399, 400, 401, 402, 403, 404, 405, 406, 407, 408, 409, 410, 411, 412, 413, 414, 415, 416, 417, 418, 419, 420, 421, 422, 423, 424, 425, 426, 427, 428, 429, 430, 431, 432, 433, 434, 435, 436, 437, 438, 439, 440, 441, 442, 443, 444, 445, 446, 447, 448, 449, 450, 451, 452, 453, 454, 455, 456, 457, 458, 459, 460, 461, 462, 463, 464, 465, 466, 467, 468, 469, 470, 471, 472, 473, 474, 475, 476, 477, 478, 479, 480, 481, 482, 483, 484, 485, 486, 487, 488, 489, 490, 491, 492, 493, 494, 495, 496, 497, 498, 499, 500, 501, 502, 503, 504, 505, 506, 507, 508, 509, 510, 511, 512, 513, 514, 515, 516, 517, 518, 519, 520, 521, 522, 523, 524, 525, 526, 527, 528, 529, 530, 531, 532, 533, 534, 535, 536, 537, 538, 539, 540, 541, 542, 543, 544, 545, 546, 547, 548, 549, 550, 551, 552, 553, 554, 555, 556, 557, 558, 559, 560, 561, 562, 563, 564, 565, 566, 567, 568, 569, 570, 571, 572, 573, 574, 575, 576, 577, 578, 579, 580, 581, 582, 583, 584, 585, 586, 587, 588, 589, 590, 591, 592, 593, 594, 595, 596, 597, 598, 599, 600, 601, 602, 603, 604, 605, 606, 607, 608, 609, 610, 611, 612, 613, 614, 615, 616, 617, 618, 619, 620, 621, 622, 623, 624, 625, 626, 627, 628, 629, 630, 631, 632, 633, 634, 635, 636, 637, 638, 639, 640, 641, 642, 643, 644, 645, 646, 647, 648, 649, 650, 651, 652, 653, 654, 655, 656, 657, 658, 659, 660, 661, 662, 663, 664, 665, 666, 667, 668, 669, 670, 671, 672, 673, 674, 675, 676, 677, 678, 679, 680, 681, 682, 683, 684, 685, 686, 687, 688, 689, 690, 691, 692, 693, 694, 695, 696, 697, 698, 699, 700, 701, 702, 703, 704, 705, 706, 707, 708, 709, 710, 711, 712, 713, 714, 715, 716, 717, 718, 719, 720, 721, 722, 723, 724, 725, 726, 727, 728, 729, 730, 731, 732, 733, 734, 735, 736, 737, 738, 739, 740, 741, 742, 743, 744, 745, 746, 747, 748, 749, 750, 751, 752, 753, 754, 755, 756, 757, 758, 759, 760, 761, 762, 763, 764, 765, 766, 767, 768, 769, 770, 771, 772, 773, 774, 775, 776, 777, 778, 779, 780, 781, 782, 783, 784, 785, 786, 787, 788, 789, 790, 791, 792, 793, 794, 795, 796, 797, 798, 799, 800, 801, 802, 803, 804, 805, 806, 807, 808, 809, 810, 811, 812, 813, 814, 815, 816, 817, 818, 819, 820, 821, 822, 823, 824, 825, 826, 827, 828, 829, 830, 831, 832, 833, 834, 835, 836, 837, 838, 839, 840, 841, 842, 843, 844, 845, 846, 847, 848, 849, 850, 851, 852, 853, 854, 855, 856, 857, 858, 859, 860, 861, 862, 863, 864, 865, 866, 867, 868, 869, 870, 871, 872, 873, 874, 875, 876, 877, 878, 879, 880, 881, 882, 883, 884, 885, 886, 887, 888, 889, 890, 891, 892, 893, 894, 895, 896, 897, 898, 899, 900, 901, 902, 903, 904, 905, 906, 907, 908, 909, 910, 911, 912, 913, 914, 915, 916, 917, 918, 919, 920, 921, 922, 923, 924, 925, 926, 927, 928, 929, 930, 931, 932, 933, 934, 935, 936, 937, 938, 939, 940, 941, 942, 943, 944, 945, 946, 947, 948, 949, 950, 951, 952, 953, 954, 955, 956, 957, 958, 959, 960, 961, 962, 963, 964, 965, 966, 967, 968, 969, 970, 971, 972, 973, 974, 975, 976, 977, 978, 979, 980, 981, 982, 983, 984, 985, 986, 987, 988, 989, 990, 991, 992, 993, 994, 995, 996, 997, 998, 999, 1000.

PIRTEA, TH.; SPAC", P.

A method of determining penicillin in finished products. p. 49.

ANALELE SERIA STINTELOR NATURII. Bucuresti, Rumania. Vol. 7, no. 20, 1958.

Monthly List of East European Accessions (EPAI), LC, Vol. 8, no. 9, Sept., 1959

Uncl.

RUMANIA / Analytic Chemistry, Analysis of Inorganic Substances.

Abs Jour: Ref Zhur-Khimiya, No 18, 1958, 60622.

Author : Th. I. Pirtea, Polly Bucsan,

Inst : "C. J. Parhon" University.

Title : New Method of Microgravimetric Determination of Cobalt. Separation and Determination of Cobalt in Presence of Nickel.

Orig Pub: An. Univ. "C. J. Parhon". Ser. stiint. natur., 1957, No 16, 71-74.

Abstract: The earlier developed method of determination of Co in the form of the complex $[\text{Co}(\text{NH}_3)_6] [\text{Co}(\text{NO}_2)_6]$ (RZhKhim, 1958, 24807) was applied to the de-

Card 1/2

PIR TOUSEK, FREDERICK

Development and prospects of the mining and metallurgy of chromium in the USSR

The article is an abridgement of a paper presented at the 1978 International Conference on Chromium, held in Moscow, USSR, in 1978. The author, Frederick P. Tousek, is a metallurgical engineer and metallurgist at the U.S. Army Research Office-Durham, Durham, North Carolina. The article discusses the development and prospects of the mining and metallurgy of chromium in the USSR. It covers the following topics: 1. The role of chromium in the USSR. 2. The development of chromium mining in the USSR. 3. The development of chromium metallurgy in the USSR. 4. The prospects of chromium mining and metallurgy in the USSR. The author concludes that the USSR has a significant advantage in the production of chromium and that the development of chromium mining and metallurgy in the USSR is a high priority for the country.

The USSR is the world's largest producer of chromium. The country has a significant advantage in the production of chromium and is a major supplier of chromium to the rest of the world. The development of chromium mining and metallurgy in the USSR is a high priority for the country. The article discusses the development and prospects of the mining and metallurgy of chromium in the USSR. It covers the following topics: 1. The role of chromium in the USSR. 2. The development of chromium mining in the USSR. 3. The development of chromium metallurgy in the USSR. 4. The prospects of chromium mining and metallurgy in the USSR. The author concludes that the USSR has a significant advantage in the production of chromium and that the development of chromium mining and metallurgy in the USSR is a high priority for the country.

The output from the remaining mines in the USSR is 15% of the total production. The country has a significant advantage in the production of chromium and is a major supplier of chromium to the rest of the world. The development of chromium mining and metallurgy in the USSR is a high priority for the country. The article discusses the development and prospects of the mining and metallurgy of chromium in the USSR. It covers the following topics: 1. The role of chromium in the USSR. 2. The development of chromium mining in the USSR. 3. The development of chromium metallurgy in the USSR. 4. The prospects of chromium mining and metallurgy in the USSR. The author concludes that the USSR has a significant advantage in the production of chromium and that the development of chromium mining and metallurgy in the USSR is a high priority for the country.

L 15062-65 EWT(m)/EPF(c)/EPR/EWP(j)/T Pc-4/Pr-4/Ps-4 RPL PW/RM
 ACCESSION NR: AP4046174 S/0079/64/034/009/2910/2911

AUTHOR: Kacheyshvili, G. Ye.; Pirtskhalava, N. I.; Dzhioshvili, G. D.

TITLE: Reaction of borontrialkyls with phenylmagnesium bromide

SOURCE: Zhurnal obshchey khimii, v. 34, no. 9, 1964, 2910-2911

TOPIC TAGS: borontrialkyl, phenylmagnesium bromide, alkyl aryl organic boron compound, triphenylboron

ABSTRACT: The reaction proceeds according to the scheme

$$n \text{ C}_6\text{H}_5\text{MgBr} + \text{R}_3\text{B} \longrightarrow (\text{C}_6\text{H}_5)_n\text{R}_{3-n}\text{B}$$

 and was conducted in ether solution under nitrogen, yielding a mixture of alkyl-aryl organic boron compounds or triphenylboron. The desired products were obtained with a 40-75% yield and are described. These were phenyldiisopropyl-, diphenylisopropyl-, phenyldi-n-propyl-, phenyldiisobutyl-, diphenylisobutyl-, phenyl-di-n-butyl-, diphenyl-n-butyl, phenylisoamyl-, phenyldi-n-amyl- and triphenylboron. Orig. art. has: 2 tables and 1 formula

Card 1/2

L 16062-65

ACCESSION NR: AP4046174

ASSOCIATION: Tbilisskiy gosudarstvennyy universitet (Tbilis State University)

SUBMITTED: 01Jul63

ENCL: 00

SUB CODE: GC, OC, MT

NO REF SOV: 001

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THE

PIRTSKHALAVA, M.D.

Concerning synthetic machine translating networks for conjunctions
and particles in the Georgian language. Trudy Inst.alek.,
avtom.i telem.AN Grus. ~~222~~ 2:113-115 '61. (MIRA 14:8)
(Georgian language--Machine translating)

L 65177-65 EWT(1)/EPA(s)-2/EWT(m)/EPP(c)/EPP(j)/T/ENA(h)/ENA(c) IJP(c)/
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ACCESSION NR: AP5022150

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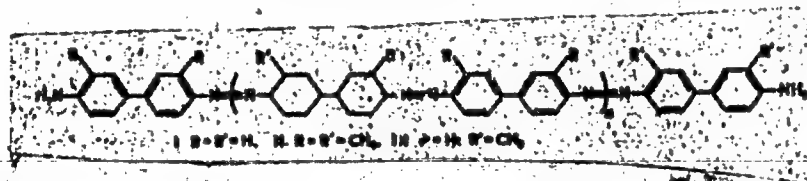
AUTHOR: Demidova, G. N.; Pirtskhalava, R. N.; Rozenshteyn, L. D.; Terpugova, M. P.;
Kotlyarevskiy, I. I.

TITLE: Absorption spectra, electrical conductivity, and photoconductivity of poly-
azopolyarenes

SOURCE: Elektrokimiya, v. 1, no. 9, 1965, 1145-1149

TOPIC TAGS: organic semiconductor, organic photoconductor, photoconductivity, elec-
tric conductivity, conjugated polymer

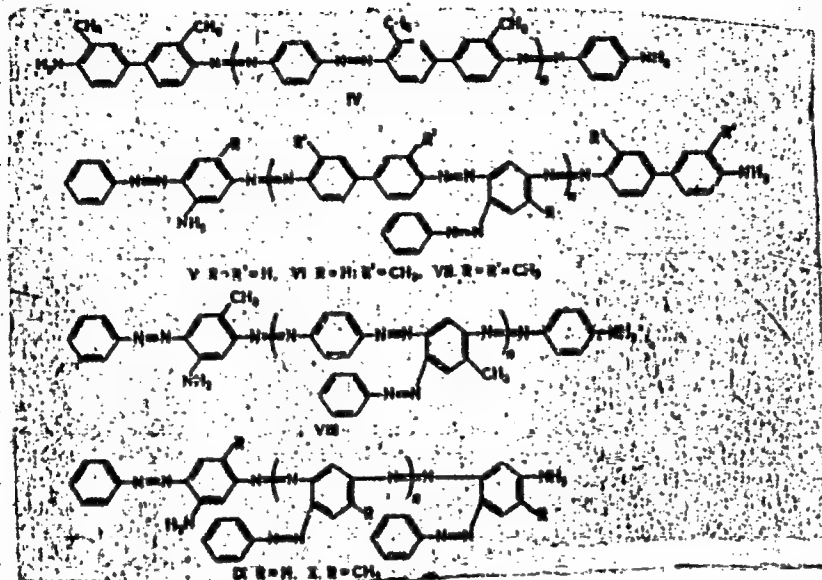
ABSTRACT: Electrical conductivity, photoconductivity and absorption spectra have
been measured for a number of polyazopolyarenes:



Card 1/6

L 65177-65

ACCESSION NR: AP5022150



Card 2/6

L 65177-65

ACCESSION NR: AP5022150

6

The materials were prepared by oxidative polycondensation or copolycondensation of aromatic diamines in the presence of cuprous chloride catalyst. The study of these compounds was prompted by the varying character of conjugation among them. The properties of the compounds are shown in Table 1 of the Enclosure. Electrical measurements were carried out with molded specimens. The temperature dependence of electrical conductivity obeyed an exponential law. For some of the compounds the $\log \sigma$ versus $1/T$ curves showed a break and hence two values of E and σ_0 are given for them in Table 1. Photoconduction was studied with film specimens. In the field range used (up to 5×10^4 v/cm) the photocurrent obeyed Ohm's law. For most films the lux-ampere characteristic was linear. The photocurrent was an exponential function of temperature. The spectral regions of photoconductivity corresponded to spectral regions of optical absorption. Comparison of absolute values of photocurrent for the polymers revealed a sharp drop in photocurrent on going to compounds with decreasing conjugated chain length. Orig. art. has: 6 figures, 1 table, and 5 formulas. [SM]

ASSOCIATION: Institut poluprovodnikov Akademii nauk SSSR (Institute of Semiconductors, Academy of Sciences SSSR); Institut khimicheskoy kinetiki i goreniya Sibirskogo otdeleniya Akademii nauk SSSR (Institute of Chemical Kinetics and Combustion, Siberian Department, Academy of Sciences SSSR)

Card 3/6

L 65177-65

ACCESSION NR: AP5022150

SUBMITTED: 09Feb65

ENCL: 02

SUB CODE: MT, GC

NO REF BOV: 006

OTHER: 000

ATD PRESS: 4089

Card 4/6

L 65177-65

ACCESSION NR: AP5022150

ENCLOSURE: 01

Table 1. Properties of polyazopolyarenes

Diamines	Polymer	mp	Solubility	Electrical conductivity data			
				E_1 , eV	σ_{01} , $\text{ohm}^{-1}\text{cm}^{-1}$	E_2 , eV	σ_{02} , $\text{ohm}^{-1}\text{cm}^{-1}$
Benzidine	I	Does not melt until 500°C	Partial in dimethylformamide	0.65	$1.13 \cdot 10^{-3}$	0.78	2.9
o-Toluidine	II	"	Partial in acetone, benzene, chloroform, dioxane			1.4	$3.65 \cdot 10^3$
o-Toluidine benzidine	III	288°C	Partial in chloroform, dioxane	0.59	$7.2 \cdot 10^{-7}$	1.6	$1.82 \cdot 10^{11}$
o-Toluidine p-phenylene diamine	IV	249°C	Partial in acetone, chloroform, benzene			2.1	$1.15 \cdot 10^{13}$
Benzidine, chrysoidane	V	352°C	Partial in chloroform, acetone			2.35	$1.15 \cdot 10^{13}$

Card 5/6

L 65177-65

ACCESSION NR: AP5022150

ENCLOSURE: 02

Table 1. Properties of polyanilines (Cont.)

Diamines	Polymer	mp	Solubility	Electrical conductivity data			
				$\bar{\epsilon}_1$, ev	$\bar{\epsilon}_2$, cm ⁻¹	$\bar{\epsilon}_1$, ev	$\bar{\epsilon}_2$, cm ⁻¹
o-Toluidine, chrysoidene	VI	286C	Partial in chloroform, acetone	0.84 1.156	$1.1 \cdot 10^{-6}$	2.1	$1.8 \cdot 10^{11}$
o-Toluidine, methylchry- soidene	VII	212- 217	Partial in benzene, ace- tone, chloro- form, dimethyl- formamide			2.95	$3 \cdot 10^{11}$
p-Phenylene- diamine	VIII	209- 215	Partial in benzene, ace- tone, chloro- form, dimethyl- formamide			3.1	$2 \cdot 10^{12}$
Chrysoidene	IX	276- 278	In benzene, cumen, ace- tone, chloro- form			3.5	$1.1 \cdot 10^{11}$
Methylchry- soidene	X	152	In benzene, cumen, ace- tone, chloro- form			3.14	$6.2 \cdot 10^{11}$

Card 8/6

L 62793-65

ACCESSION NR: AP5018457

ENT(m)/EPT(c)/EPR/EMP(1)/I/EMA(c)

PC-L/PT-L/PU-L NW/JAJ/EM

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621.315.592:547

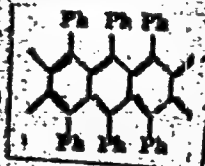
AUTHOR: Davydov, B. E.; Daidova, G. M.; Nasirov, F. M.; Pirtakhalava, R. M.;
Rozenshteyn, L. D.

TITLE: Synthesis of polydiphenyldiacetylenes and their electrical and physical properties

SOURCE: Elektrokimiya, v. 1, no. 7, 1965, 876-880

TOPIC TAGS: polymerization synthesis, acetylene, thermal stability, catalysis, photoelectric current

ABSTRACT: The article is concerned with the investigation of the properties of thermally polymerized diphenyldiacetylene, having the following structure



Card 1/5

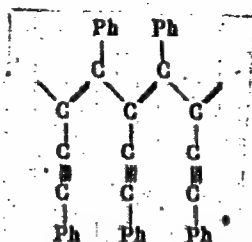
L 62793-65
ACCESSION NR: AP501B*57

The fact that all hydrogen atoms of the main chain are replaced by phenyl groups accounts for its solubility in chloroform, benzene, ether, dioxane, dimethylformamide and for its high thermal stability. It may be heated as high as 500° C without any significant decomposition. The polymer is a nonfusible dark brown substance. When it is deposited from solution on a substrate it forms a relatively strong film. The kinetic polymerization curves at different temperatures are shown in Fig. 1 of the Enclosure. The cryoscopic data indicates that the molecular weight of the polymer is 1100. Experiments with polymer films show that upon interaction with oxygen under the influence of light this polymer does not have a tendency to form inner peroxides. Diphenyldiacetylene was also polymerized catalytically using a complex catalyst, produced during interaction of triethylaluminum and vanadyl acetylacetonate. The polymer obtained was also soluble in benzene, chloroform and other organic solvents. The thermal stability of this polymer was somewhat lower than in the polymers produced by thermal initiation. The lower stability is explained by the presence of the following structures in the chain

Card 2/6

L 62793-65

ACCESSION NR: AP5018457



The photoelectric conductivity of thermal polymers is observed in the region where they absorb light. The spectral dependence of the photoelectric current reduced to the same amount of energy incident on the specimen is in good agreement with the absorption spectrum (Fig. 2 of the Enclosure). Catalytic polymers displayed no photoelectric conductivity. An attempt was made to measure the dark current, but it was possible to record it only when the field strength exceeded $2 \cdot 10^4$ V/cm. Orig. art. has: 5 figures.

ASSOCIATION: Institut neftekhimicheskogo sinteza Akademii nauk SSSR (Petrochemical Synthesis Institute Academy of Sciences SSSR)

Card - 3/6

L 62793-65
ACCESSION NR: AP5018457

Institut poluprovodnikov Akademii nauk SSSR (Institute of Semiconductors Academy of Sciences SSSR)

SUBMITTED: 09Feb65

ENCL: 02

SUB CODE: OC, EM

NO REF SOV: 005

OTHER: 000

Card 4/6

L 62793-65

ACCESSION NR: AP5018457

ENCLOSURE: 01

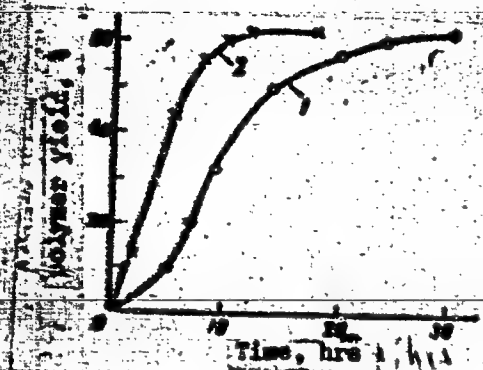


Fig. 1. Kinetic curves for thermal polymerization of diphenyldiacetylene at 106° C (1) and 133° C (2)

Cond 5/8

L 62173-45

DOCUMENT NO: AP5018457

ENCLOSURE: 02



Fig. 2. Spectral absorption curves and photoelectric current per unit incident energy, of polydiphenylacetylene, heated at 150° C for 3 hrs 45 min. The curves on the right are given on smaller scale.

KUTATELADZE, K.S.; PIPTSEHALAVA, Ye.A.

Investigating Borzhomi andesites in connection with the manufacture of dark glazes. Soob. AN Gruz.SSR 21 no.6:673-676 D '58.

(MIRA 12:4)

1. AN Gruz.SSR, Institut metallurgii. Predstavleno chlenom-korrespondentom Akademii F.N. Tavadze.
(Borzhomi--Andesites) (Glazes)

USSR/Cultivated Plants. Subtropical. Tropical.

M-8

Abs Jour: Ref. Zhur-Biologiya, No 5, 1958. 20532.

Author : S. Kh. Pirtskhalayshvili

Inst : The All-Union Scientific Research Institute for Tea and Subtropical Crops.

Title : Contribution to the Problem of Better Stocks for Citrus Crops. (K voprosu o luchshikh podboyakh dlya tsitrusovykh kul'tur).

Orig Pub: Byul. Vses. n.-i. in-ta zhaya i subtrop. kul'tur, 1956, No 4, 72-80.

Abstract: Preliminary data is given on the tests conducted from 1950 to 1956 to study stocks for citrus fruit cultures under field conditions. As stocks one tried the trifoliata, bitter orange, Kaba lemon, seedlings of the Novogruzinskiy lemon, seedlings of the Mestny orange, Natsau-Mikan, Tsitrus

Card : 1/4

USSR/Cultivated Plants. Subtropical. Tropical.

Abs Jour: Ref Zhur-Biologiya, No 5, 1958, 20532.

M-8

Yunos, The Grapefruit Duncan, Tsitrantz, Kinkan; as grafts, the Novogruzinskiy lemon, seedlings of the Novogruzinskiy lemon, seedlings of the Monakello lemon, the Washington Navel orange, seedlings of the Mestniy orange, the Unshiu mandarin. Eyes of the Novogruzinskiy lemon adapted themselves almost uniformly well to all stocks (95-97%); eyes inoculated into the Kaba lemon (96%) and minimal in the bitter orange and trifoliata (53-55%); eyes of the Washington Navel orange had maximum adaptability on seedlings of the Novogruzinskiy lemon (91%). Eyes of the Unshiu mandarin showed maximum compatability when they were inoculated in seedlings of the Novogruzinskiy lemon and Kaba lemon (77-78%), then on the trifoliata (65%) and least on the bitter orange (45%). The most vigorous growth in all

Card : 2/4

USSR/Cultivated Plants. Subtropical. Tropical.

Abs Jour: Ref Zhur-Biologiya, No 5, 1958, 20532.

CIA-RDP86-00513R001341020002-8

M-8

stocks observed was witnessed in the Novogruzinskiy lemon, then come the Washington Navel orange and Monakello lemon. The Unshiu mandarin orange displayed vigorous growth on all stocks. Among the powerfully growing stocks are cited the seedlings of the Novogruzinskiy lemon, then the Kaba lemon, the bitter orange, the grapefruit, orange seedlings; the weakest growth is in the trifoliata. Of all the stocks tested the trifoliata was the most suitable stock for open ground citrus cultures, inasmuch as it promoted increased frost resistance and earlier fruit bearing of the citrus crops and guaranteed a steady growth in the grafts. Low growing trees are easier to protect in cold weather and it is more convenient to take care of them. On this stock all citrus varieties show satisfactory eye adaptability.

Card : 3/4

S/137/60/000/011/034/043
A006/A001

Translation from: Referativnyy zhurnal, Metallurgiya, 1960, No. 11, p. 251,
27245

AUTHORS: Tavadze, P.N., Pirtskhalayshvili, V.A.

TITLE: Investigating the Structure of Cast-Iron of the Chrome-Manganese System

PERIODICAL: Dokl. Nauchno-proizv. konferentsii mashinostroiteley i priboro-
stroiteley, Leningrad, Sudpromgiz, 1959, pp. 184 - 194

TEXT: It was revealed that in alloys of the Fe-Cr-Mn system at a C con-
tent of 0.12%, C widens considerably the range of alloys with an austenite base,
shifting it toward the side of higher Cr concentrations. At C 2.2 - 2.4% this
range includes alloys containing > 25% Cr. The austenite of Cr-Mn-cast irons
with Mn 15 - 20%, Cr from 0 to 6 - 8%, decomposes partially into perlite starting
from 675 - 700°C; austenite, however, does not undergo perlite decomposition. ✓
In low-chromium alloys, in the range of austenite-base Cr-Mn-system of cast-irons,
a part of Mn is bound in (Fe,Mn,Cr)₃C carbides. In high chromium alloys, Cr is

Card 1/2

S/137/60/000/011/034/043
A006/A001

Investigating the Structure of Cast-Iron of the Chrome-Manganese System

mainly bound in carbides and its ferrite-forming capacity vanishes. Si affects the stability of the austenite of the Cr-Mn system. In the Cr-Mn system of cast irons C is bound in carbides rich in Cr and Mn. Free C is present in the structure only in alloys with a low Cr and Mn content. There are 18 references. ✓

A.S.

Translator's note: This is the full translation of the original Russian abstract.

Card 2/2

ZEDGINIDZE, Ye.N.; PIRTSKHALAVA, Ye.A.; MANULASHVILI, N.K.; BAGATUROVA,
I.A.

Studying laterite clays of the Tsetskhlauri deposit. Soob.AN Gruz.
SSR 25 no.5:539-542 N '60. (MIRA 14:1)

1. Akademiya nauk GruzSSR, Institut prikladnoy khimii i elektro-
khimii, Tbilisi. Predstavleno akademikom R.I.Agladze.
(Kobuleti District--Laterite)

CHARGEYSHVILI, A.K., prof.; PIRTSKHALAYSHVILI, V.A.

Report on the activity of the Georgian Scientific Medical
Society of Otorhinolaryngologists for 1962. Vest. oto-rin.
25 no.4:104-106 JI-Ag '63. (MIRA 17:1)

1. Pre 'datel' Gruzinskogo nauchnogo meditsinskogo obshchestva
otorinolaringologov (for Chargeyshvili). 2. Sekretar'
Gruzinskogo nauchnogo meditsinskogo obshchestva otorinolarin-
gololov (for Pirtskhalayshvili).

TAVADZE, P.N.; PIRTSEHALAYSHVILI, V.A.

Effect of high carbon content in the austenite field of the system
iron-chromium-manganese. Soob. AN Gruz.SSR 21 no.6:727-733 D '58.
(MIRA 12:4)

1. AN GruzSSR, Institut metallurgii, Tbilisi. 2. Chlen-korrespon-
dent AN GruzSSR (for Tavadze).
(Iron alloys)

CHARGEYSHVILI, A.K., prof. PIRTSEKHALAYSHVILI, V.A.

Report on the work of the Georgian Society of Otorinolaryngologists
in 1956. Vest.oto-rin. 19 no.4:120-122 J1-Ag '57. (MIRA 10:11)

1. Predsedatel' Gruzinskogo respublikanskogo nauchnogo obshchestva
oto-rino-laringologov (for Chargeyshvili). 2. Sekretar' Gruzinskogo
respublikanskogo nauchnogo obshchestva oto-rino-laringologov (for
Pirtsekhlayshvili)
(OTORHINOLARYNGOLOGISTS)

PIRUMOV, Kh.M.

Data on the development of control of parasitic diseases in the
Armenia S.S.R. Med.paras. i paras.bol. 26 no.5:578-581 8-0 '57.
(MIRA 11:2)

1. Iz Instituta epidemiologii i gigiyeny Ministerstva zdrevo-
okhraneniya Arayenskoj SSR.

(PARASITIC DISEASES, prev. & control
in Armenia (Rus))

Piritea, Despin

chem A new gravimetric method for the rapid determination of copper. Despin Piritea, Beatrice Ruzutu, and Suzana Zelig. *Acad. rep. Populare Romine. Studii cercetari chim.* 3, 233-6 (1955) (French summary). The method is based on the pptn. of Cu^{2+} as $[Cupr.]_2(CIO)_4$ with $NaCIO_4$ (I) in the presence of pyridine (II). Add II dropwise to 20-30 ml. of the soln. until the blue color does not change and a slight odor of II is perceptible; add an excess of I (2-3 g.); stir, and let the ppt. stand until the supernatant is clear and

colorless. (If the soln. is still bluish, add more I and agitate again). Filter the soln. through a filter crucible A; wash several times with an eq. soln. satd. with I contg. 2 drops of II/50 ml.; next wash 5-8 times with 2-ml. portions of a soln. contg. abs. EtOH (3 ml.), abs. Et₂O (7 ml.), II (3 drops), and I (0.01 g.), and finally wash 4-5 times with 2-3-ml. portions of abs. Et₂O. Dry for 15 min. in a vacuum desiccator (to const. wt.) and filter. The accuracy of the method is within $\pm 0.4\%$. Gary Gerard

PIRTEA, DESPINA

3

New gravimetric methods for the rapid determination of nickel, cobalt, and cadmium. Despins Pirtea, Gh. Dumitrescu, and J. Meleucu. *Acad. Rep. Populara Romania, Studii Cercetari chim.* 3, 217-22 (1955) (French summary). The methods are based on the pptn. and direct weighing of the insol. complexes $[X(py)_4]Cr_2O_4$, where $X = Ni, Co$, and Cd . To 35-50 ml. soln. contg. Ni , add pyridine (I) dropwise until the blue color is not intensified by further addn. Add an excess of an aq. soln. satd. with $K_2Cr_2O_7$ (II), heat cautiously to boiling while agitating continuously, and then boil for a few more min. until the ppt. has become cryst. (evidenced by rapid settling). Add 2 more drops of I and 1 ml. of II; cool to room temp.; filter through a filter crucible; wash several times with a soln. contg. II (3 ml.), I (0.5 ml.), and H_2O (100 ml.); and then wash 5-6 times with 2-ml. portions of abs. $EtOH$ contg. 50 drops of I in 10 ml. Apply a relatively high vacuum, and wash 3-4 times with 2-ml. portions of abs. H_2O . Dry in a vacuum desiccator at room temp. and weigh. Co and Cd are detd. similarly. All 3 methods are accurate to within $\pm 0.2\%$. (Gary Ceram)

Chem

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PIRTEA, DESPINA

USSR

✓ 100. Separation and spectrophotometric determination of copper in the presence of iron or aluminum or both. PIRTEA, DESPINA and DUMITRU, PIRTEA (Comm. Acad. R.P. Romania, 1983, 5-12-83, 51-52). Referats Zh. Khim., 1984, Abstr. No. 31,000. — Copper is pptd. as $CuPy_2(CNS)_2$ (Specu and Dick, Z. anal. Chem., 1987, 71, 184) from a soln. containing tartrate, which prevents the pptn. of Fe and Al. Fe and Al are pptd. from the filtrate by an ethanolic soln. of 8-hydroxyquinoline. R. Hayes

ALTEA, E.; ATT, E.; ELIO, E.

New gravimetric method for rapid determination of copper. p. 233.

Academia Republicii Populare Romine. STUDII DE METALURGIE

Iucuresti. Vol. 3, no. 3/4, July/Dec. 1955

So. East European Accessions List Vol. 5, No. 9 September, 1966

PIRTHA, I.; TROF, A.

A new microgravimetric method for the determination of zinc.

P. 435 (REVISTA DE CHIMIE) (Bucuresti, Rumania) Vol. 8, No. 6, June 1957

SO: Monthly Index of East European Accessions (MEAI) LC Vol. 7, No. 5, 1958

NIETLA, T.

Theorghe Spacu; an obituary. p. 155. Academia Republicii Populare
Romane. STUDII SI CERCETARI DE LIMBA. Bucuresti. Vol. 3, no. 3/4, July/
Dec. 1955.

So. East European Acquisitions list Vol. 5, No. 2 September, 1956

PIRTEA, T.

A new method for the separation and gravimetric dosage of zinc. p. 183.
Vol. 2, No. 3/4, July/Dec. 1954, Bucuresti, Rumania.

SOURCE: East European Accessions List, Library of Congress, Vol. 5, No. 10,
October 1956.

PIPTA, T.

"A New Method for the gravimetric determination of teryllium. P/O
BULATIN STIINTIFIC, Vol. 3, No.2-L, Apr./Dec. Bucuresti, Romania.

SC: Monthly List of East European Accessions, L.C.Vol. 2, No. 11, Nov, 1964.
Uncl.

PIRTEA, T.

rapid precise process for separation and gravimetric
quantitative analysis of zinc in the presence of iron
and aluminum. p. 459.
ACADEMIA REPUBLICII ROMANE. Rumania. Vol. 5,
no. 5, May 1955.

SOURCE: REAL LC Vol. 1, No. 11, August 1956

ANALYSIS/Analysis of Inorganic Substances

G-2

Abs Jour: J. of Chem. Ed., 1967, 44, 1043

Author : J. Spacu, M. Hirta

Inst : U. C. Parham University.

Title : New method of quantitative determination of Mercury in the presence of Iron and Aluminum.

Orig Pub: Anal. Chim. Acta, 1967, 81, 1043

Abstract: Hg^{2+} ions are precipitated as Hg_2Cl_2 precipitate after Fe^{3+} and Al^{3+} have been combined in salicylate complexes. Fe and Al are determined in the filtrate, using a known method.

Card 1/1

- 19 -

Pirtea, Th. I.

~~new gravimetric methods for the determination of thorium, aluminum, beryllium, and zinc and their separation from certain elements. G. Spacu and Th. I. Pirtea (Univ. "I. Parhon," Bucharest). Rev. chim., 32(2), 77, 1985, 1986, Roumaine 1, No. 2, 6-25 (1950) (in French).—In a modification of a method with mercaptobenzothiazole (I) for the detn. of Cu, Cd, Pb, Ti, Bi, and Au (C.A. 29, 7213; 30, 2875), a procedure is described by which Th, Al, Zn, and Be are estd. gravimetrically by means of the Na salt (II) of I. To 5-20 ml. of a $\text{Th}(\text{NO}_3)_3$ soln. contg. 0.01-0.1 g. Th, add 2-10 ml. of a 10% aq. soln. of II with agitation. The pptd. 1-Th (III), white crystals, is filtered, washed with 50-100 ml. of a soln. contg. 0.1-0.15 g. of II and distd. H_2O , and dried at 110-20°. The factor is 0.2676. — III can be calc'd to ThO_2 at 1100°. Al. It is detd. by a similar method as a salt of I (factor 0.051307), or by the calcination of the latter to Al_2O_3 . In the presence of Mg, Al is pptd. 1st with II, and after the pptn. of I with 10-15% HCl , Mg is detd. in the filtrate with an alc. soln. of 8-quinolinol (IV) (Berg, C.A. 21, 3850). Be. A neutral or weakly acidic Be salt soln. (5-50 ml.) contg. 0.003-0.06 g. Be is pptd. with 1-15 ml. of the soln. of II. The resulting 1-Be(II), H_2O is washed with warm 3% NH_4NO_3 soln. contg. II, dried over P_2O_5 , and calcined to the oxide. SO_4^{2-} , NO_3^- , halogen⁻, OAc^- , Na^+ , K^+ , NH_4^+ do not interfere with this detn. The sepn. of Be from Mg is carried out similarly to the estn. of Al and Mg. Be in the presence of Al, Al is pptd. with IV (Kolthoff and Sandell, C.A. 32, 3112), and Be from the filtrate with II at 60°. The Be-I is estd. as oxide. Zn. The Zn salt soln. (5-50 ml.) (pH 5-6), contg. 1-2 g. NaCl is pptd. with a 10% soln. of II (factor 0.1044). In the presence of Al and Fe, 1-2 g. of tartaric acid is added and Zn pptd. as Zn-I and calcined to ZnO . Zr. It is pptd. at pH 3.9 and estd. as ZrO_2 .~~

Distr: 4E3d

PIRTEA, T. I.

3

1900. A new rapid method for estimation of beryllium. *Chim. Ber.* 1900, 27, 115. The amphoteric character of Be, which forms easily

soluble beryllates and salts that readily hydrolyse, hinders the formation of a single stable complex. This difficulty is overcome by combining Be with a suitable complex ion, and then with a second one. This is achieved by treating soln. of Be^{2+} with conc. soln. of ammonium carbonate. The sol. beryllium complex thus formed is treated with a conc. soln. of hexa-amminocobalt chloride, which gives a very insoluble cryst. yellow ppt. of $[Be_2(CO_3)_2(OH)_2][Co(NH_3)_6]_2 \cdot 3H_2O$, which is easily washed and dried in a vacuum desiccator. The method is simple and quick (40 to 50 min.), and can be used in the presence of alkaline metals and Fe. Since the Be content of the ppt. is only 4.1%, accurate determination of small quantities is possible.

H. SARA

PM — for any

PIRTEA, T.I.

2894. New gravimetric and volumetric method for determination of silver. P. Scau and T. I. Pirta. *Rev. Chim., Bucharest*, 1968, 7 (9), 431-435. The procedure is based on the reaction of Ag⁺ with sodium nitroprusside (I), which gives a cream ppt. of Ag₂[Fe(CN)₅(NO)]₂, unaffected by light, stable, and insoluble, with mol. wt. greater than that of the usual halogen complexes. Pptn. is rapid and complete at room temp., and ppt. can be filtered immediately, and after washing can be dried in a vacuum desiccator or even in an oven at 110°. If modified the method can be used in the presence of Pb and Zn. *Gravimetric method*—To 10 to 20 ml of a neutral or acid soln. of Ag⁺ at 50° to 60° add 1 to 2 g of solid NH₄NO₃, followed by approx. 0.1 N I. A yellow-red colour of the supernatant liquid indicates complete pptn. and excess of I. Filter immediately through a sintered glass crucible, washing with NH₄NO₃ soln. (3%), water, ethanol, and ether. Dry in a vacuum desiccator and weigh. The determination takes 1 to 1.5 hr. In the presence of Pb or Zn the ppt. is washed 4 to 5 times with aq. NH₄NO₃ soln. (3%) heated to between 50° and 60°. *Volumetric method*—Since addition of I soln. to AgNO₃ soln. leads to the formation of a colloidal ppt., the determination is carried out by running AgNO₃ soln. into a known vol. of I. This gives a good end-point with or without eosin as an adsorption indicator. Results are consistently ± 0.2% high.

H. SERR

PIRTEA, Th.I.

New methods for the gravimetric determination of copper. Rev
chimie Min petr 13 no.4:234-235 Ap '62.

Pirtea, Th. L.

7000

A new method for the separation and gravimetric determination of the G, Si and Th. J. Pucci (C. I. Smith Correlation Chem., 2, 187-90 [1951] French summary). Zn can be precip. as $Zn(C_6H_5NHS)_2$ by adding an excess of slightly acidic soln. of Na salt of mercaptobenzoic thioamide to the at pH 8-10, filtering, washing, drying and drying at 115-125°. Adding a little NaCl improves the filtration. At 800-1000° gives ZnO . Calculating Gary General

PM

RUMANIA / Analytical Chemistry. Inorganic Analysis.

Abs Jour : Ref Zhur - Khimiya, No 23, 1959, No. 81928

Author : Spacu, P.; Pirtea, Th. I.

Inst : Not given

Title : Potentiometric Determination of Silver in the Presence of Other Elements

Orig Pub : An. Univ. "C. I. Parhon". Ser. stiint. natur., 1958, No 20, 55-58

Abstract : It has been determined that the method previously developed by the authors for the determination of Ag^+ by the potentiometric titration with sodium nitroprusside solution (RZ Khim, No 9, 1958, No. 24742) is also applicable when Tl^+ and most elements that are found with Ag in alloys and ores (Pb, Cd, Zn, Cu, Co, Ni, Mn, 3b) are present in the solution. The solution

Card 1/2

Gravimetric method for the determination of beryllium in the presence of other elements in alloys and minerals. Th. I. Pirtea and V. Constantinescu (Univ. "C. I. Pirtea," Bucharest, Romania). *Z. anal. Chem.* 183, 183-8 (1959).
—In the presence of (ethylenedinitrilo)tetracetic acid (I) Cu, Cd, Zn, Co, Ni, Fe, Al, Ti, Mn, Mg, Ca, Na, K, and NH_4^+ do not interfere in the pptn. of $[\text{Co}(\text{NH}_3)_4][(\text{H}_2\text{O})_2\text{Be}(\text{CO}_3)_2(\text{OH})_2 \cdot 3\text{H}_2\text{O}]$. For the detn. of up to 75 mg. Be in 5-25 ml. of weakly acid soln. add 0.5-1 g. NH_4Cl and 0.5-2 g. I (more if large amts. of Fe^{+++} are present). If a ppt. forms, add 10% NH_4OH dropwise until the ppt. dissolves. Add 2-3 g. powd. $(\text{NH}_4)_2\text{CO}_3$. Add without mixing 0.5-2 ml. of $\text{Co}(\text{NH}_3)_4\text{Cl}_2$ soln. (II) and wait for the ppt. to start forming. Add an excess of II. Dil. the soln. so that the salt concn. is 2-3%, let stand for 1.5-2 hrs., and filter into a sintered crucible. Wash with 0.2% I, 2-4 times with 3-ml. portions 60% EtOH contg. a few drops 0.2% I, and finally with EtOH and Et₂O. Dry in a vacuum desiccator.

W. O. Stone

Distr: 4E20

RUSIYA, Zaur; KHURTSILAVA, Giga; SMAKHARADZE, Kukuri; MIKAYA, Zurab;
SIRADZE, Bando; AVAZASHVILI, Guguli; PIATSKHALASHVILI, Pavle;
TATUASHVILI, Anzor

Search goes on. Sov. profsoyuzy 18 no.5:16-18 Mr '62.
(MIRA 15:3)

1. Zavod "Elektroavtomat", g. Tbilisi.
(Tiflis--Labor and laboring classes)

PIRTSEHALAISHVILI, S.Kh., kand.s.-l'skokhoz.nauk

Research work on tea cultivation in principal tea growing provinces
of China. Biol.VNIICHISK no.2:154-163 '57. (MIRA 15:5)
(China-Tea research)

I 18727-66 ENT(m)/ENA(d)/ENP(t) IJP(c) JD/JQ

ACC NR: AP6005092

SOURCE CODE: UR/ 0251/65/040/003/0685/0692

AUTHOR: Tavadze, P. M. (Academician AN GruzSSR); Pirtskhalashvili, V. A.;
Khutsishvili, M. L.

ORG: Georgian Institute of Metallurgy (Gruzinskiy institut metallurgii)

TITLE: Effect of chromium on the structure and properties of nitrogen-containing
austenitic chromium-manganese and chromium-manganese-nickel steels 27
18

SOURCE: AN GruzSSR. Soobshcheniya, v. 40, no. 3, 1965, 685-692

TOPIC TAGS: chromium, austenitic steel, nitrogen, plastic deformation, annealing,
chromium steel, manganese steel

ABSTRACT: Specimens of specially prepared alloys containing different proportions of
technically pure Fe, electrolytic Cr (13.89-21.60%) and Mn (11.72-12.20%) and
nitrided electrolytic Cr¹ and Mn¹ (with ~6% N) were hot-worked (annealing at 1200°C for
5 hr + immediate water quenching or cooling at room temperature over 24 hr) were
tested for microhardness, hardness, electric resistance and deformation resistance.
Microstructural examination and phase identification were based on the use of various
etching agents. Findings: Cr-Mn steels containing 16% Mn, 16-18% Cr and 0.40-0.50% N
display the highest deformation resistance at 700°C under a stress of 15 kg/mm². If
the Cr content deviates from the 16-18% range, deformation resistance decreases

Cord 1/2

L 18727-66

ACC NR: AP6005092

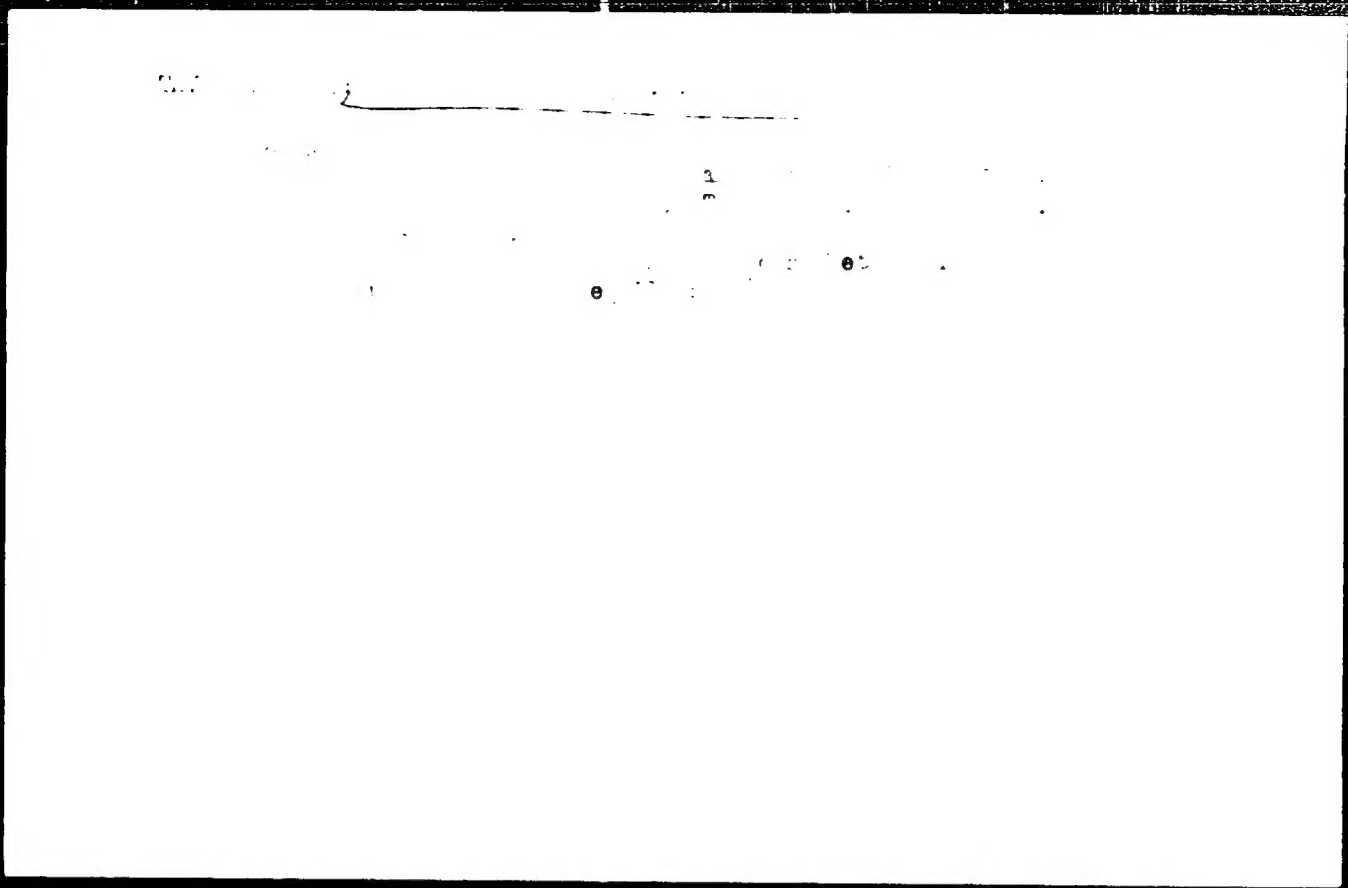
sharply owing to the decrease in the Cr concentration of the γ -solid solution, appearance of porosity in ingots, and formation of a ferritic component in the structure. Cr-Mn steels containing up to 15.50% Cr display a higher deformation resistance when in quenched state compared with annealed state, whereas for the steels containing from 15.50 to 21.50% Cr this picture is reversed. Such an effect of hot working is apparently attributable to the difference in the rates of the aging process in the steels with a Cr content below and above 15.50%. The hardness and microhardness of the investigated Cr-Mn and Cr-Mn-Ni (~3% Ni) steels in quenched state are markedly higher than in annealed state. This may be due to the special character of the aging of these steels or to the low-temperature metastable transformation. The change in the deformation resistance of Cr-Mn-Ni steel at 700°C as a function of the concentration of Cr indicates that deformation resistance sharply increases in the presence of Cr concentrations of up to 17% but does not change appreciably above that limit. The presence of N in austenitic Cr-Mn and Cr-Mn-Ni steels in an amount not below its solubility limit in the γ -solid solution and not above its solubility limit in the melts of these steels markedly enhances their deformation resistance under conditions of prolonged exposure to high temperatures and loads. Orig. art. has: 2 tables, 4 figures.

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Aluminum salts from terra rossa. S. M. Ponomarev, M. V. Ponomareva, and M. V. Ponomareva. *Trudy Inst. Khim., Akad. Nauk Gruz. S.S.R.* 11, 67-70 (1968) (in Georgian; Russian summary); *Referat. Zhur., Khim.* 1968, No. 40002. — The 3-stage extn. of Al_2O_3 burnt red earth of Western Georgia was studied. The earth was treated with either H_2SO_4 or HCl . The salts of Al and Fe obtained in the 1st stage were sep'd. by adding untreated red earth and removing from soln. the major part of Fe salts by means of hydrolysis followed by coagulation of Fe hydroxides. The weak acid formed in the hydrolysis process ext'd. Al salts with only a small admixt. of Fe salts. The HCl process yielded purer Al salts than did the H_2SO_4 process. For final purification of Al salts they were crystd. from soln. as $AlNH_4$ salts which were subsequently converted into pure Al_2O_3 and $(NH_4)_2SO_4$. In the first stage of extn. of sesquioxides from red earth a highly dispersible amorphous SiO_2 was obtained. M. Hosh.

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